



Aalto University

# Thermal radiation

*Summer School on Fire Dynamics Modeling 2017*

*Jülich Supercomputing Centre (JSC)*

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# Objectives

1. Know the phenomena and parameters affecting Radiation Transport Equation (RTE).
2. Can explain the RTE discretization principles in FVM/DOM models.
3. Can investigate the sensitivity of results on the numerical parameters.
4. Can specify FDS inputs for radiation model.
5. Can modify the FDS verification tests.
6. *Understand the difference between gray gas and spectrally resolved calculations.*
7. *Can describe the challenges of predicting the radiative fraction.*
8. *Know the relationship between gas species/soot concentration and the calculated absorption coefficient.*
9. *Can identify the role of radiation modelling in validation.*
10. *Can tune the time-related solver parameters for verification tests.*

# Contents

1	Part 1. Radiation fundamentals
2	Part 2. Radiation transport equation
3	Derivation of discrete form of Finite Volume Method
4	Exercise 1. Radiation in a box
5	Part 3. Radiation properties
6	Source term modelling in FDS
7	Exercise 2. Hottel's charts

# Recommended reading

Howell, J.R., Pinar Menguc, M., Siegel, R. Thermal Radiation Exchange, (6. ed), CRC Press, 2015.

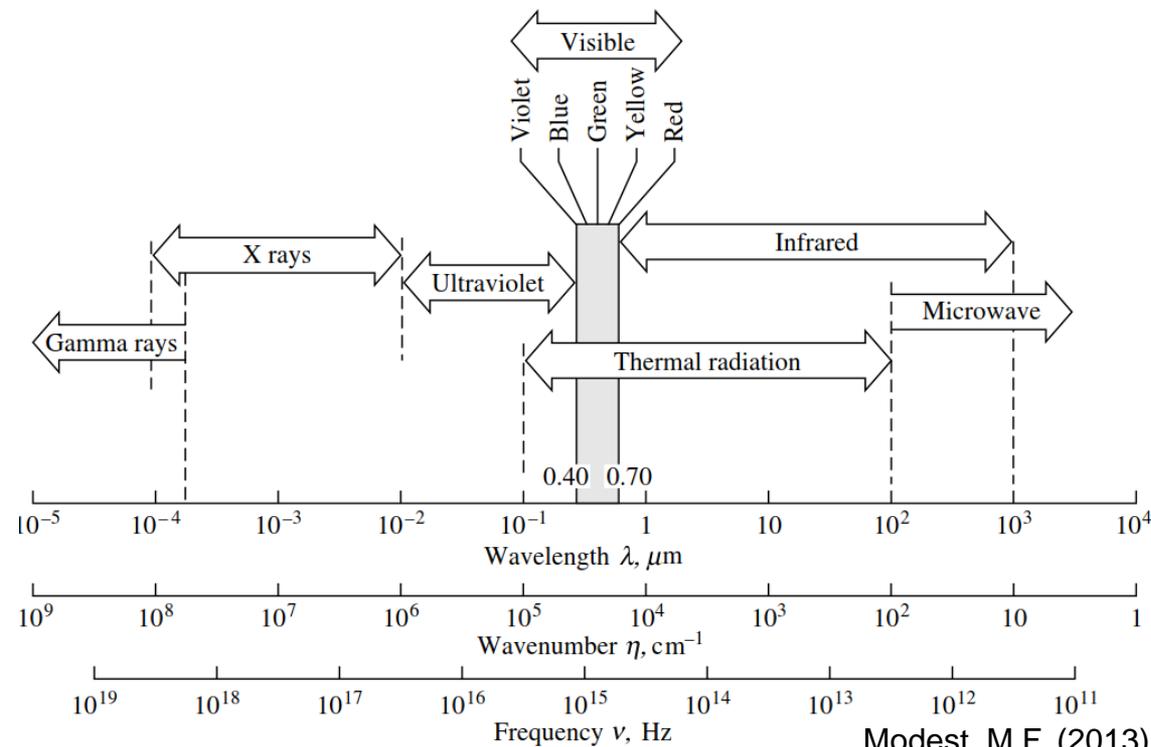
Michael F. Modest. Radiative Heat Transfer (3. ed) Academic Press. (1993, 2003, 2013)



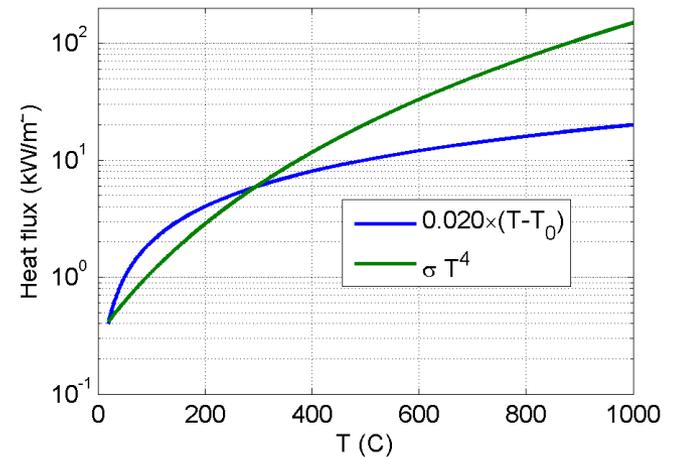
# Part 1: Fundamentals

# Thermal radiation

- Thermal radiation is electromagnetic radiation (at infrared (IR) range).
- What radiates?
- Radiation dominates fire heat transfer when
  - Flame diameter  $> \sim 0,3$  m
  - Temperature  $> 400$  °C

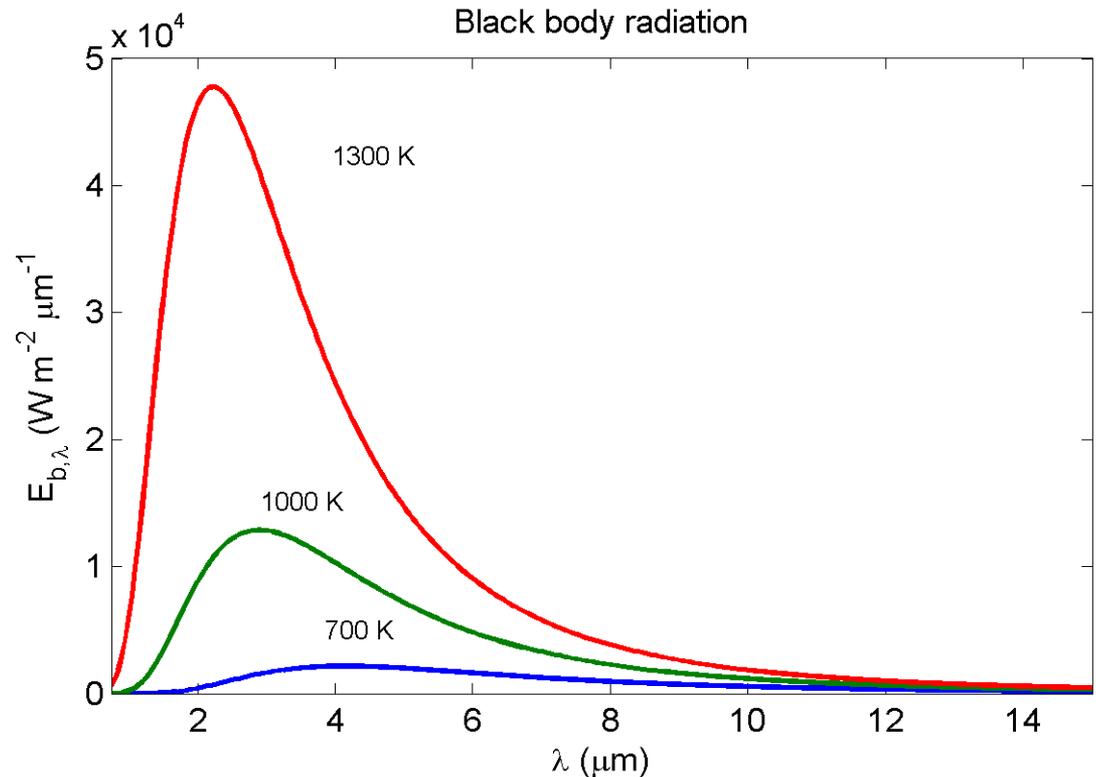


Modest, M.F. (2013)



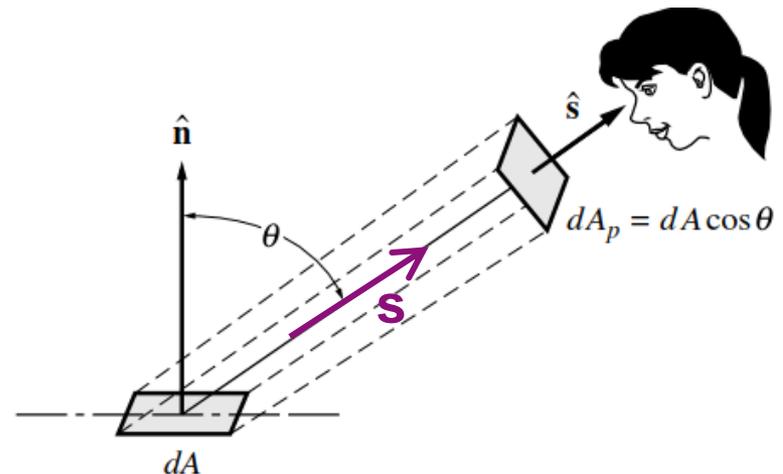
# Black body -radiation

- Energy density depends on the wavelength and the temperature of the radiating body.
- Qualitatively
  - 550 °C: reddish
  - 900 °C: bright red
  - 1100 °C: orange
  - 1500 °C: white



# Definitions

- **Spectral emissive power**  $E_\lambda$  (or  $E_\nu$ )  
≡ radiative energy / time / surface area / wavelength ( $\text{W m}^{-2} \mu\text{m}^{-1}$ )
- **Total emissive power**  $E$  ( $\text{W m}^{-2}$ )
- **Spectral intensity (or radiance)**  $I_\lambda$   
≡ radiative energy / time / area normal to rays / solid angle / wavelength ( $\text{W m}^{-2} \text{sr}^{-1} \mu\text{m}^{-1}$ )
- **Total intensity** ( $\text{W m}^{-2} \text{sr}^{-1}$ )
- Blackbody emissive power  $E_b$
- Blackbody emissive intensity  $I_b$
- Direction vector  $\mathbf{s}$



# Stefan-Boltzmann law

- Power  $P$  emitted by black surface with area  $A$
- $\sigma$  = Stefan-Boltzmann constant  $5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$
- Real surfaces are not black, but emit less than the ideal surfaces.

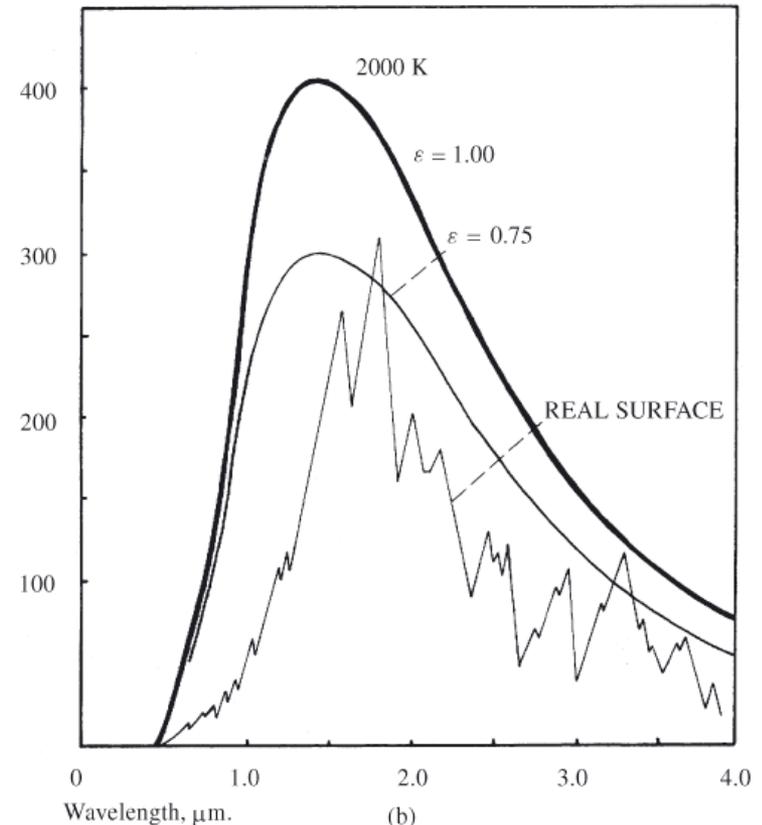
- Surface emissivity (and absorptivity)

$$\varepsilon(\lambda) = \frac{E_{\lambda}}{E_{b,\lambda}}$$

- For *gray bodies*  $\varepsilon = \text{constant} \Rightarrow$

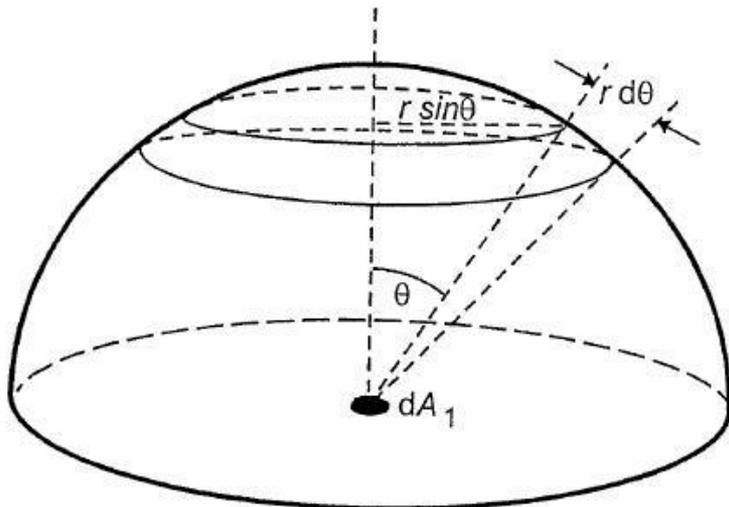
$$\dot{q}'' = \varepsilon \sigma T^4$$

$$\frac{P}{A} = E_b = \dot{q}_b'' = \sigma T^4$$



# Intensity vs. emissive power

- Black body radiates at same intensity to all directions  $I(\theta) = I_n$ .
- The projected area of the radiating surface decreases when the observer goes away from the surface normal.
- Surfaces like this are called Lambert radiators or *diffuse radiators*, and they obey the cosine law



$$E_{\lambda,b}(\theta)d\lambda = I_{\lambda,b} \cos \theta d\lambda = I_{\lambda,n} \cos \theta d\lambda$$

$$\Rightarrow E_b = \int_0^{\infty} E_{\lambda,b} d\lambda = I_b \int_0^{\pi/2} \int_0^{2\pi} \cos \theta \sin \theta d\theta d\phi = I_b \cdot \pi$$

$$\Rightarrow I_b = E_b / \pi$$

# Attenuation in participating media

Intensity of monochromatic radiation attenuates as

$$dI_\lambda = -\kappa_\lambda C I_\lambda dx$$

where  $\kappa_\lambda$  is specific absorption coefficient and  $C$  is concentration.

Integrating in domain  $x = 0 \dots L$  we get  $I_{\lambda L} = I_{\lambda 0} \exp(-\kappa_\lambda C L)$

Monochromatic absorptivity  $a_\lambda = \frac{I_{\lambda 0} - I_{\lambda L}}{I_{\lambda 0}} = 1 - \exp(-\kappa_\lambda C L)$

A world map is shown in the background, with a semi-transparent grey rectangular box centered over the Atlantic and Indian Oceans. The text 'Part 2. Radiation transport' is written in a bold, purple font within this box. The map uses a color gradient from light yellow to dark brown to represent different regions or data points.

## Part 2. Radiation transport

# Radiation Transport Equation (RTE)

General equation for radiation intensity at point  $\mathbf{x}$  to direction  $\mathbf{s}$

$$\frac{1}{c} \frac{\partial I_\lambda(\mathbf{x}, \mathbf{s}, t)}{\partial t} + \mathbf{s} \cdot \nabla I_\lambda(\mathbf{x}, \mathbf{s}) = \underbrace{-\kappa(\mathbf{x}, \lambda) I_\lambda(\mathbf{x}, \mathbf{s})}_{\text{Energy loss by absorption}} - \underbrace{\sigma_s(\mathbf{x}, \lambda) I_\lambda(\mathbf{x}, \mathbf{s})}_{\text{Energy loss by scattering}} +$$
  
$$\underbrace{B(\mathbf{x}, \lambda)}_{\text{Emission source term}} + \underbrace{\frac{\sigma_s(\mathbf{x}, \lambda)}{4\pi} \int_{4\pi} \Phi(\mathbf{s}', \mathbf{s}) I_\lambda(\mathbf{x}, \mathbf{s}') d\mathbf{s}'}_{\text{In-scattering term}}$$

# Spectrally integrated RTE

If the spectral details are not important, we can integrate RTE over a finite number of spectral bands (ignoring scattering for simplicity)

$$\mathbf{s} \cdot \nabla I_n(\mathbf{x}, \mathbf{s}) = B_n(\mathbf{x}) - \kappa_n(\mathbf{x}) I_n(\mathbf{x}, \mathbf{s}), \quad n = 1 \dots N$$

$$B_n(\mathbf{x}) = \kappa_n(\mathbf{x}) I_{b,n}(\mathbf{x})$$

$$I_{b,n}(\mathbf{x}) = F_n \frac{\sigma [T(\mathbf{x})]^4}{\pi}$$

In typical fire simulation,  $N = 1$

# Radiation and the energy equation

Radiant heat flux vector

$$\dot{\mathbf{q}}_r''(\mathbf{x}) = \int_{4\pi} \mathbf{s}' I(\mathbf{x}, \mathbf{s}') \, d\mathbf{s}'$$

Radiative source term in energy equation

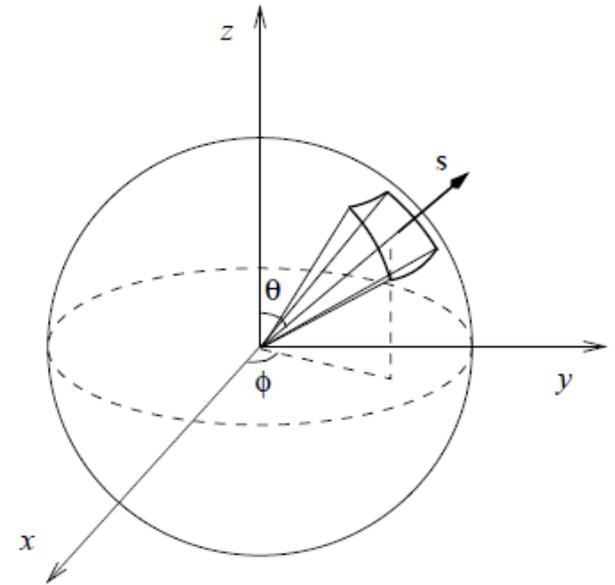
$$-\nabla \cdot \dot{\mathbf{q}}_r''(\mathbf{x})(\text{gas}) = \kappa(\mathbf{x}) [U(\mathbf{x}) - 4\pi I_b(\mathbf{x})] \quad ; \quad U(\mathbf{x}) = \int_{4\pi} I(\mathbf{x}, \mathbf{s}') \, d\mathbf{s}'$$

Boundary condition for intensity

$$I_w(\mathbf{s}) = \frac{\varepsilon \sigma T_w^4}{\pi} + \frac{1 - \varepsilon}{\pi} \int_{\mathbf{s}' \cdot \mathbf{n}_w < 0} I_w(\mathbf{s}') |\mathbf{s}' \cdot \mathbf{n}_w| \, d\mathbf{s}'$$

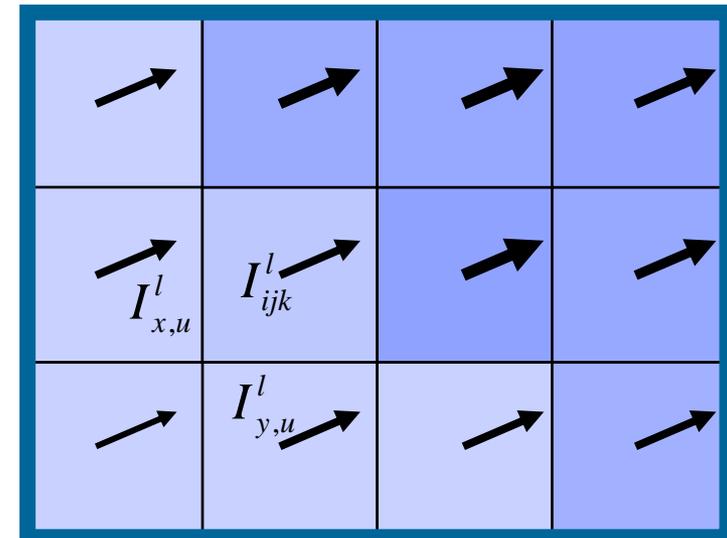
# Finite Volume Method for Radiation 1

- Discretized solid angle
  - Elevation angle  $\theta$  divided into  $N_\theta$  parts (bands)
  - Each elevation band divided into  $N_\phi(\theta)$  parts over azimuthal angle.
  - Symmetrical over co-ordinate axis planes.
- Explicit solution in CFD mesh.



## FDS User's Guide:

```
NUMBER_RADIATION_ANGLES  
TIME_STEP_INCREMENT  
ANGLE_INCREMENT
```



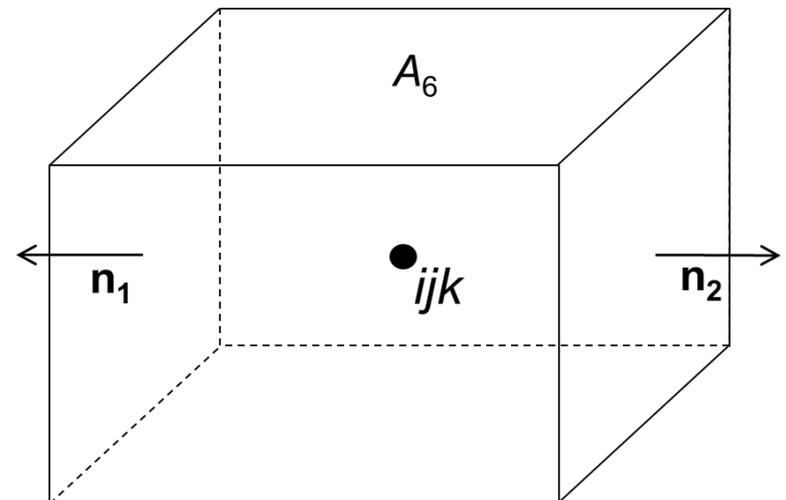
# Finite Volume Method for Radiation 2

$$\int_{\delta\Omega^l} \int_{V_{ijk}} \mathbf{s}' \cdot \nabla I(\mathbf{x}', \mathbf{s}') d\mathbf{x}' ds' = \int_{\delta\Omega^l} \int_{V_{ijk}} \kappa(\mathbf{x}') [I_b(\mathbf{x}') - I(\mathbf{x}', \mathbf{s}')] d\mathbf{x}' ds' \quad (6.14)$$

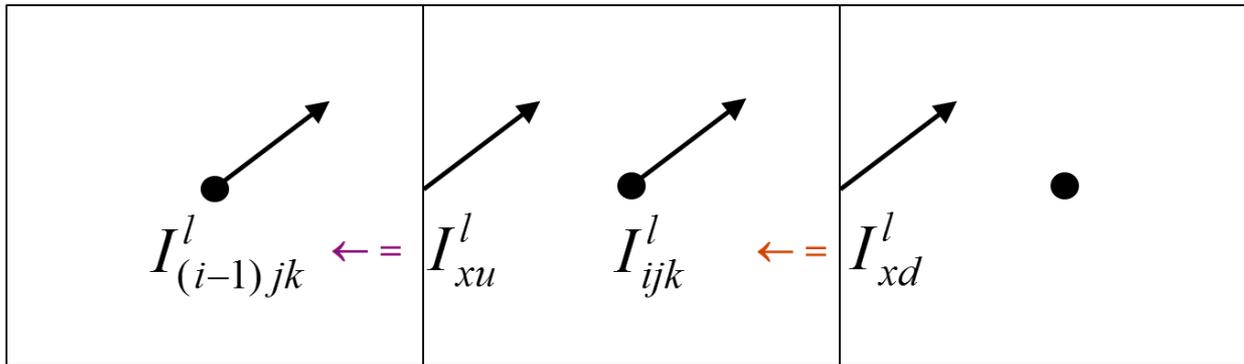
$$\int_{\delta\Omega^l} \int_{A_{ijk}} (\mathbf{s}' \cdot \mathbf{n}') I(\mathbf{x}', \mathbf{s}') d\mathbf{n}' ds' = \int_{\delta\Omega^l} \int_{V_{ijk}} \kappa(\mathbf{x}') [I_b(\mathbf{x}') - I(\mathbf{x}', \mathbf{s}')] d\mathbf{x}' ds' \quad (6.15)$$

$$\sum_{m=1}^6 A_m I_m^l \int_{\Omega^l} (\mathbf{s}' \cdot \mathbf{n}_m) ds' = \kappa_{ijk} [I_{b,ijk} - I_{ijk}^l] V_{ijk} \delta\Omega^l \quad (6.16)$$

Equation (6.16) is a discrete equation for intensity  $I_{ijk}^l$ . Solving (6.16) requires geometry data,  $\kappa_{ijk}$ , and face intensities  $I_m^l$



# Finite Volume Method for Radiation 3



- In the FVM solver of FDS, the cell face intensities are approximated using step scheme (first order upwind)
- Define angular integrals  $D_m^l = \int_{\Omega'} (\mathbf{s}' \cdot \mathbf{n}_m) ds'$
- The discrete equation for  $I_{ijk}^l$  now becomes

$$I_{xu}^l = I_{(i-1)jk}^l$$

$$A_x I_{xu}^l D_{xu}^l + A_x I_{ijk}^l D_{xd}^l +$$

$$A_y I_{yu}^l D_{yu}^l + A_y I_{ijk}^l D_{yd}^l +$$

$$A_z I_{zu}^l D_{zu}^l + A_z I_{ijk}^l D_{zd}^l$$

$$= \kappa_{ijk} I_{b,ijk} V_{ijk} \delta\Omega^l - \kappa_{ijk} I_{ijk}^l V_{ijk} \delta\Omega^l$$

(6.20)

# Finite Volume Method for Radiation 4

- Combining  $I_{ijk}^l$  terms gives

$$a_{ijk}^l I_{ijk}^l = a_x^l I_{xu}^l + a_y^l I_{yu}^l + a_z^l I_{zu}^l + b_{ijk}^l$$

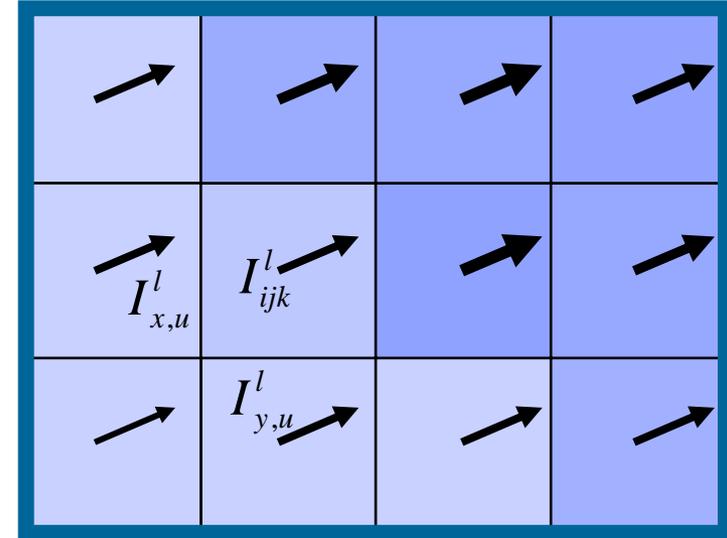
$$a_{ijk}^l = A_x |D_x^l| + A_y |D_y^l| + A_z |D_z^l| + \kappa_{ijk} V_{ijk} \delta\Omega^l$$

$$a_x^l = A_x |D_x^l|$$

$$a_y^l = A_y |D_y^l|$$

$$a_z^l = A_z |D_z^l|$$

$$b_{ijk}^l = \kappa_{ijk} I_{b,ijk} V_{ijk} \delta\Omega^l$$

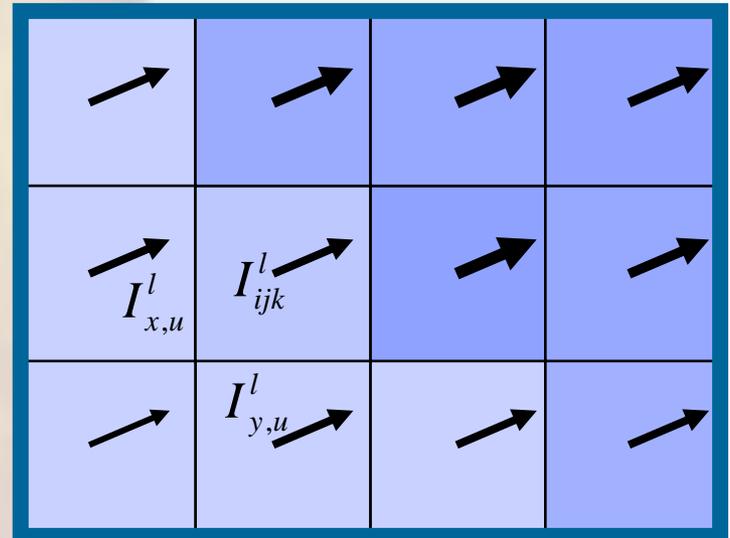


Pair discussion:

How to choose which corner to start the "sweep" from?

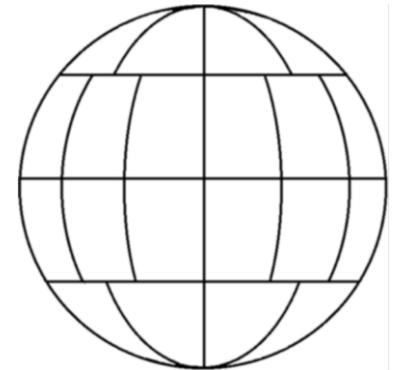
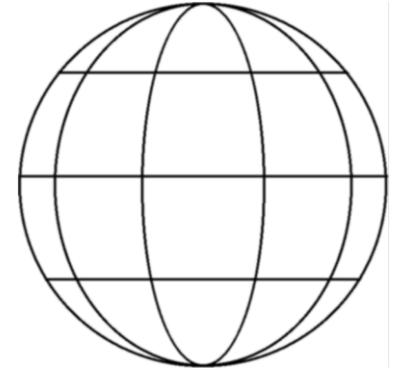
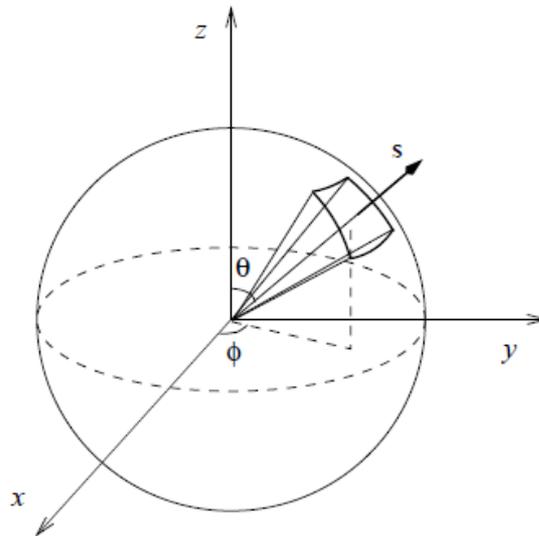
Assume you know integrals

$$D_m^l = \int_{\Omega^l} (\mathbf{s}' \cdot \mathbf{n}_m) ds'$$



# Angular discretization schemes

- Classical Discrete Ordinates Method (DOM) schemes with  $N_\phi \times N_\theta$  control angles.
  - Number of azimuthal ( $\phi$ ) bands
  - Number of polar ( $\theta$ ) bands
- More elaborate schemes naturally exist.
- FDS strategy: Equal control angles in all directions. Default number of control angles is 104.



# FVM vs. DOM

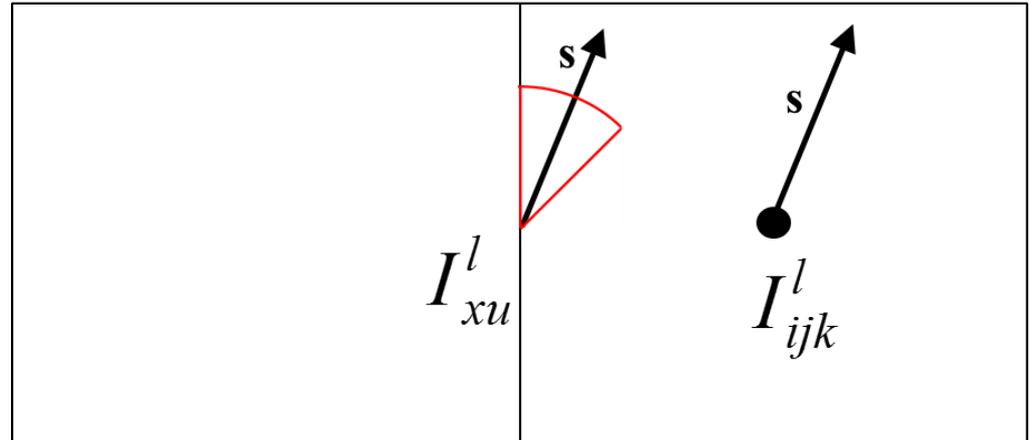
- The main difference between DOM and FVM is the treatment of angular integral

$$D_m^l = \int_{\Omega'} (\mathbf{s}' \cdot \mathbf{n}_m) d\mathbf{s}'$$

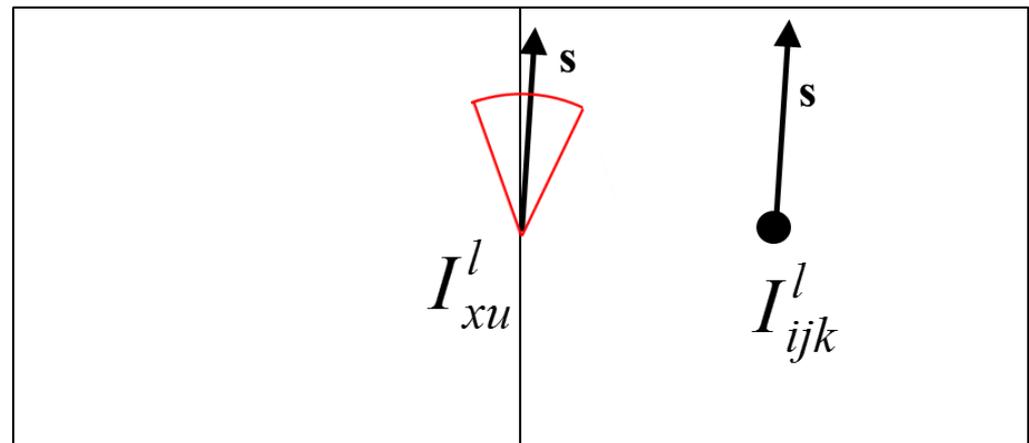
- In FVM, these integrals are calculated analytically and tabulated for the solver.
  - Exact, energy-conserving.
  - Difficult if control angles overlap with co-ordinate axes.
- In DOM, these integrals are *approximated* using *weighting sets* ( $S_n$ ).

# FVM angular discretization challenges

No overlap  
(overhang)  $\Rightarrow$   
 $D_m$  is easy to calculate

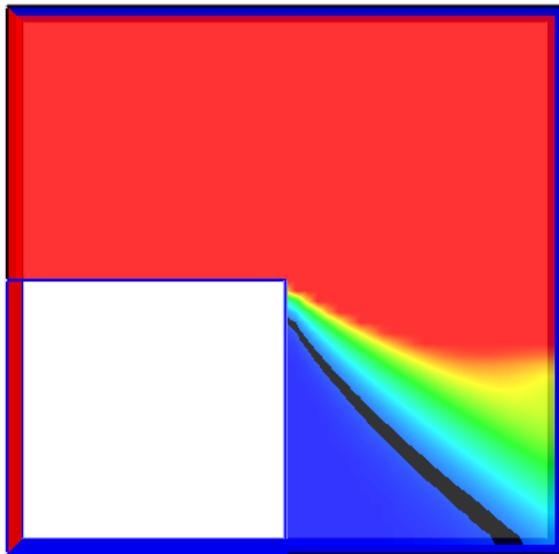


Control angle  $\delta\Omega$  overlaps  
with cell boundary  
 $\Rightarrow$  Only part of energy  
within  $\delta\Omega$  contributes to  
the flux at  $ijk$ .  
 $\Rightarrow D_m$  is difficult to  
calculate.

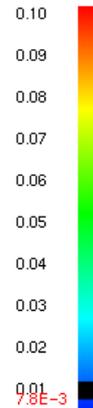


# Challenges of numerical methods

## Numerical diffusion

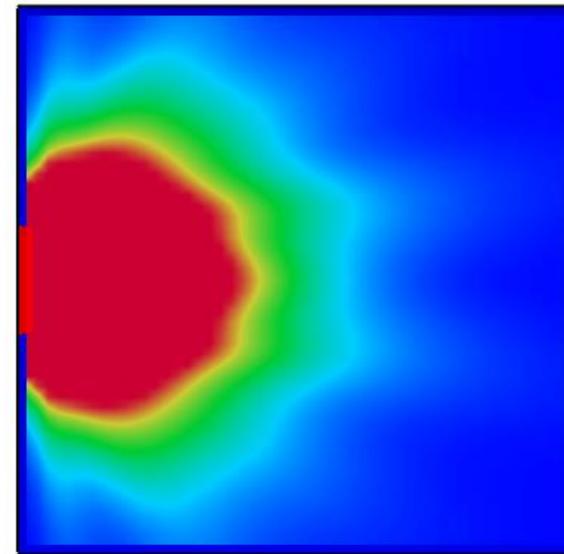


Slice  
U  
kW/m<sup>2</sup>

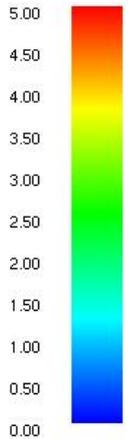


Time: 0.002

## Ray effect



Slice  
U  
kW/m<sup>2</sup>  
 $\times 10^{-2}$



Time: 0.003

A world map is shown in the background, with a semi-transparent grey rectangular box overlaid on the Americas. The map uses a color scale where warmer colors (red, orange, yellow) represent higher values and cooler colors (blue, green) represent lower values. The grey box is centered over North and South America.

# Exercise 1. Radiation box

# Exercise 1: Radiation inside a box

## 8.2 Radiation inside a Box (`radiation_in_a_box`)

This verification case tests the computation of three-dimensional configuration factor  $\Phi$  inside a cube box with one hot wall and five cold (0 K) walls. An overview of the test geometry is shown in Fig. 8.2. The

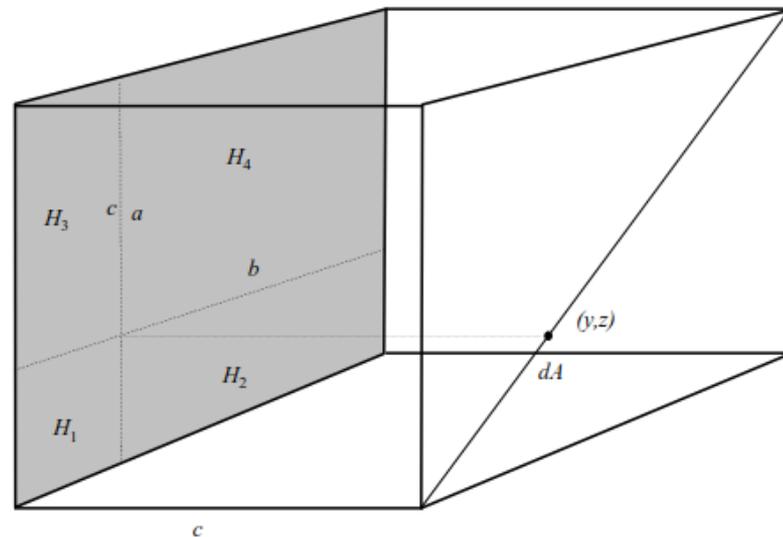


Figure 8.2: Radiation in a box geometry.

# Exercise 1: Verification results

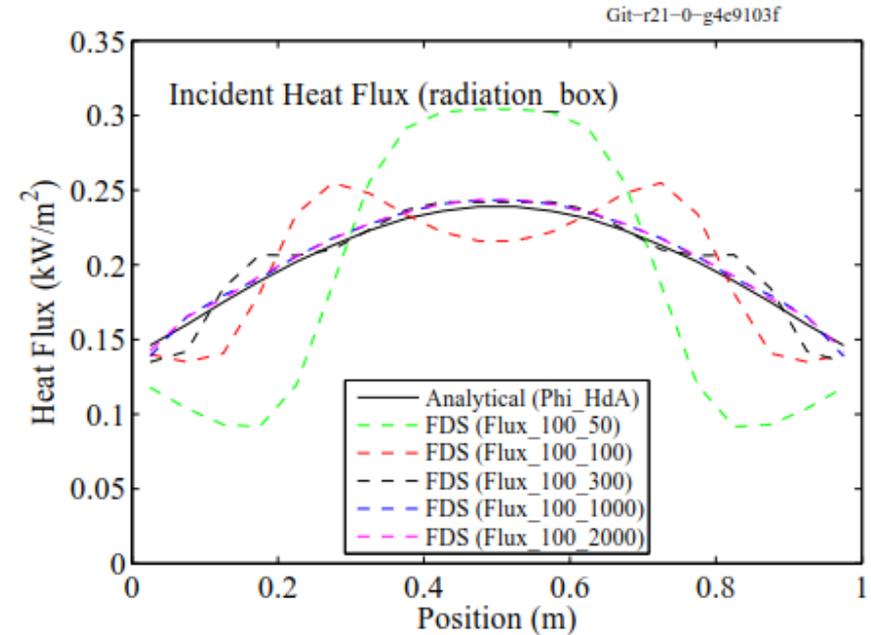
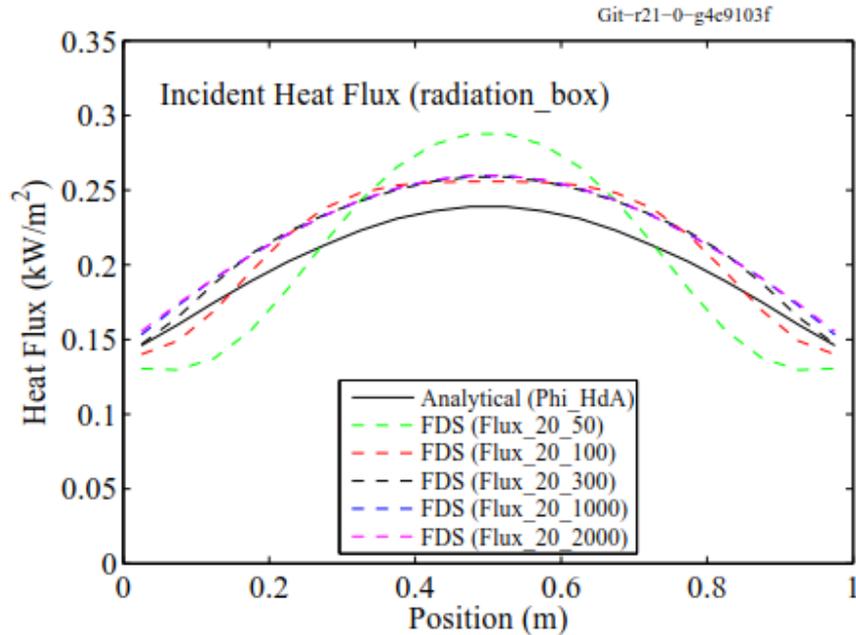
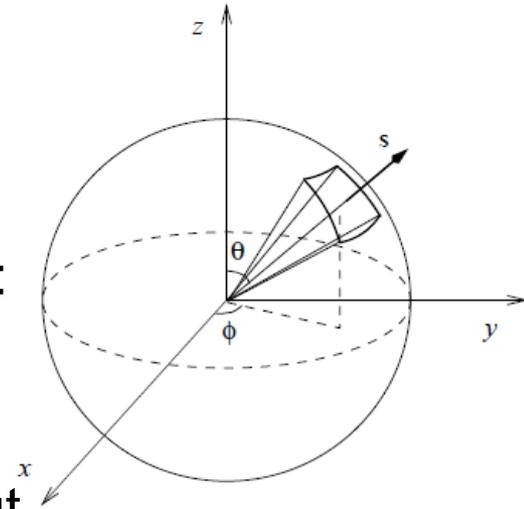


Figure 8.3: Incident heat flux.

# Exercise 1:A Solver parameters

1. Run the radiation\_box\_\_50\_\_100.fds
2. What is the purpose of the following input lines:  

```
&RADI NUMBER_RADIATION_ANGLES=100  
TIME_STEP_INCREMENT = 1  
ANGLE_INCREMENT = 1 /
```
3. Look at the 'Radiation Model Information' in .out file. How many radiation control angles are used and how are they distributed? Roughly how many degrees do the control angles span in  $\theta$ - and  $\phi$ -directions?
4. Using the results of the FDS Verification guide, discuss the effect of the angular and spatial resolutions on the heat flux distribution of the opposite wall.



# Exercise 1:B Angular discretization

**Investigate if the predicted heat fluxes are invariant to the choice of primary axis direction X, Y or Z.**

## **Steps:**

1. Modify the radiation\_box\_\_50\_\_100.fds by changing the emitting hot surface to the YMIN or ZMIN boundary.
2. Name the cases as radiation\_box\_\_50\_\_100\_y and radiation\_box\_\_50\_\_100\_z.
3. Plot the heat fluxes on the diagonal in all three cases (X,Y,Z) and compare against the exact values using plot\_box.py (edit the script to choose what to plot.)
4. Discuss the findings with your pair.

A world map is shown in the background, with a semi-transparent grey rectangular box overlaid on the Americas. The map uses a color gradient from light yellow to dark brown to represent different regions or data points. The text "Part 3. Radiation properties" is centered within the grey box in a bold, purple font.

## Part 3. Radiation properties

# Radiation Transport Equation (RTE)

$$\mathbf{s} \cdot \nabla I_\lambda(\mathbf{x}, \mathbf{s}) = \underbrace{-\kappa(\mathbf{x}, \lambda) I_\lambda(\mathbf{x}, \mathbf{s})}_{\text{Energy loss by absorption}} - \underbrace{\sigma_s(\mathbf{x}, \lambda) I_\lambda(\mathbf{x}, \mathbf{s})}_{\text{Energy loss by scattering}} +$$

$$\underbrace{B(\mathbf{x}, \lambda)}_{\text{Emission source term}} + \underbrace{\frac{\sigma_s(\mathbf{x}, \lambda)}{4\pi} \int_{4\pi} \Phi(\mathbf{s}', \mathbf{s}) I_\lambda(\mathbf{x}, \mathbf{s}') ds'}_{\text{In-scattering term}}$$

Integrated over a wavelength band  $n$  (usually  $N=1$ )

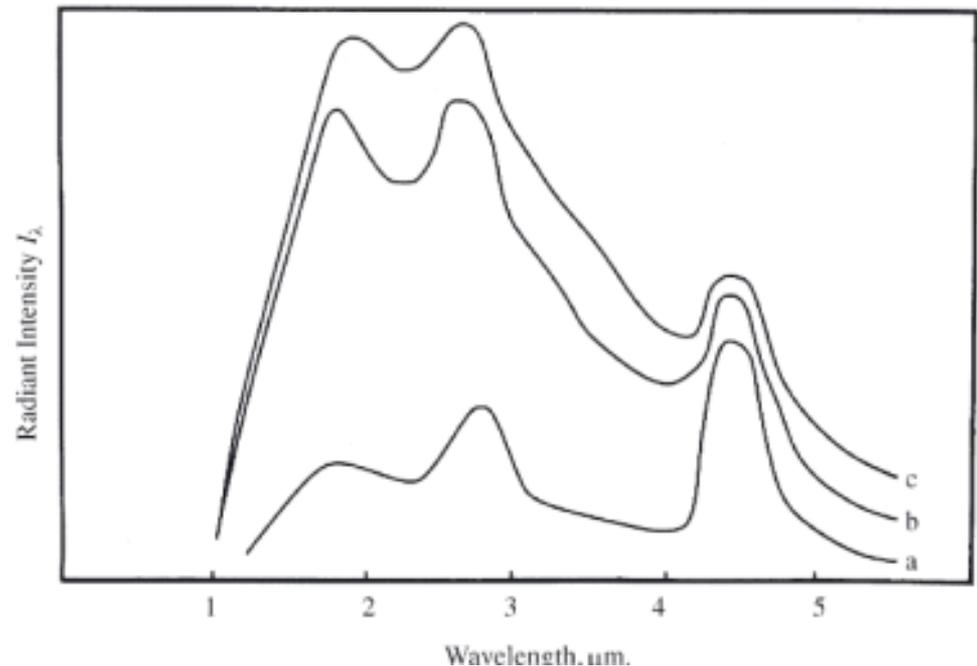
$$\mathbf{s} \cdot \nabla I_n(\mathbf{x}, \mathbf{s}) = B_n(\mathbf{x}) - \kappa_n(\mathbf{x}) I_n(\mathbf{x}, \mathbf{s}), \quad n = 1 \dots N$$

$$B_n(\mathbf{x}) = \kappa_n(\mathbf{x}) I_{b,n}(\mathbf{x})$$

# Flame radiation

- Flame radiation is emitted by soot and multi-atomic gases ( $\text{H}_2\text{O}$ ,  $\text{CO}_2$ ,  $\text{CO}$ ,  $\text{HCl}$ ,  $\text{HCN}$ ,  $\text{NO}_3$ ,...).
- $\text{O}_2$  and  $\text{N}_2$  do not emit or absorb radiation.
- Soot radiation spectrum is continuous and very close to black body radiation.
- Gas radiation spectra are very discontinuous.

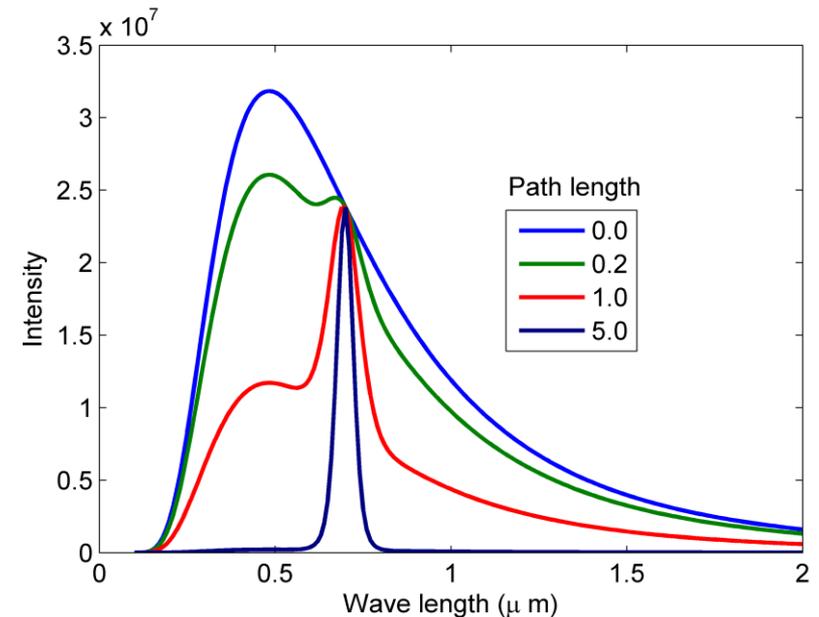
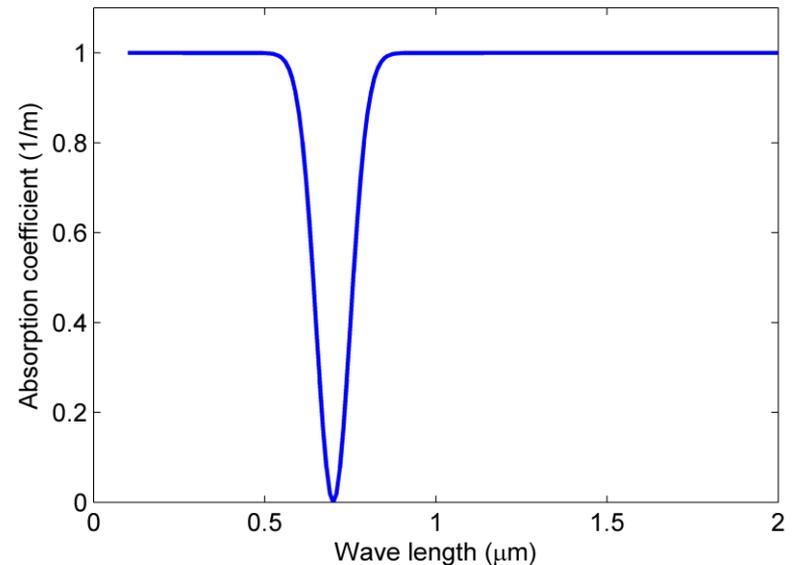
Wood flame radiation for different flame thicknesses (Drysdale / Hägglund & Peterson 1976)

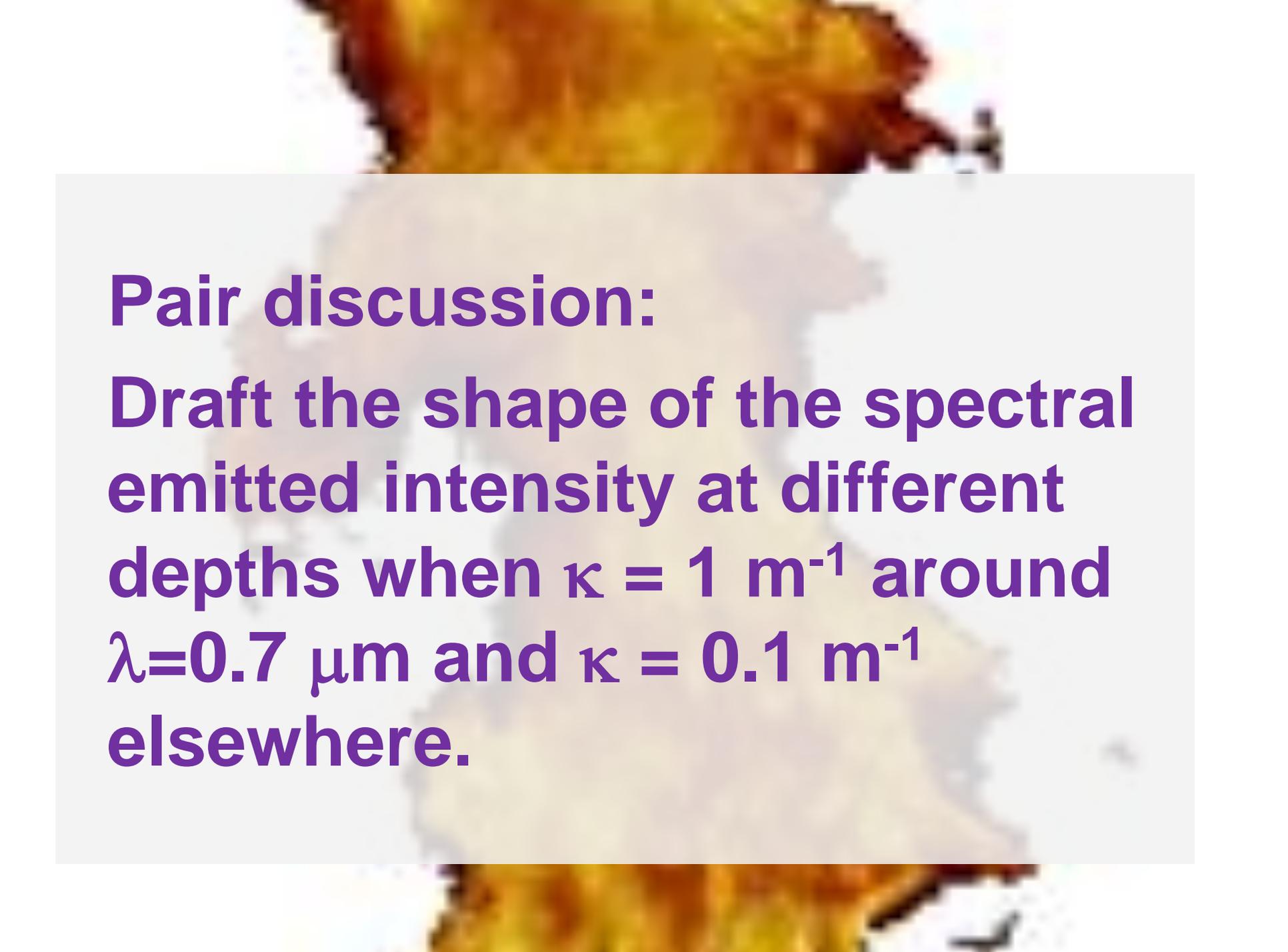


# Absorption by discontinuous spectrum

- Assume a black body radiation entering gas with spectral absorption coefficient that is constant, except for  $\lambda=0.7 \mu\text{m}$ , where it goes to zero.
- Intensity changes now differently at different wavelengths, according to equation

$$I_{\lambda}(L) = I_{\lambda,0} \exp(-\kappa_{\lambda}L)$$



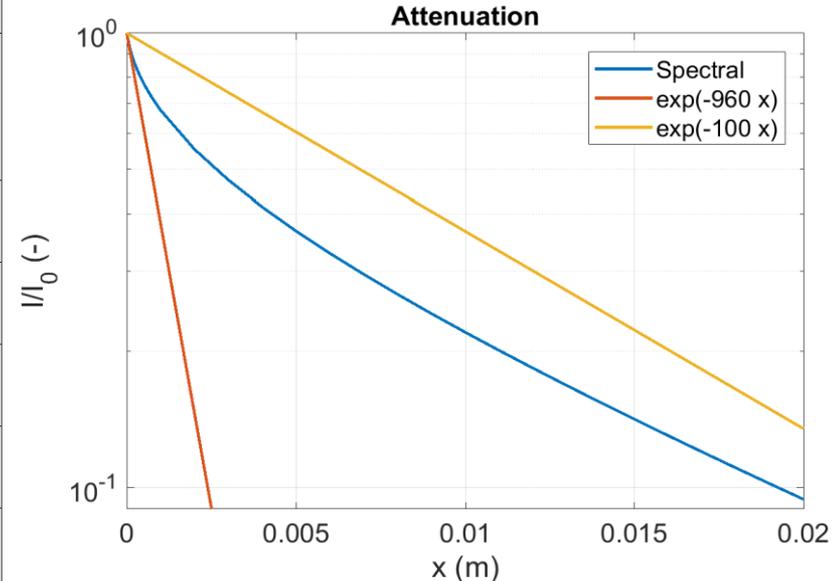
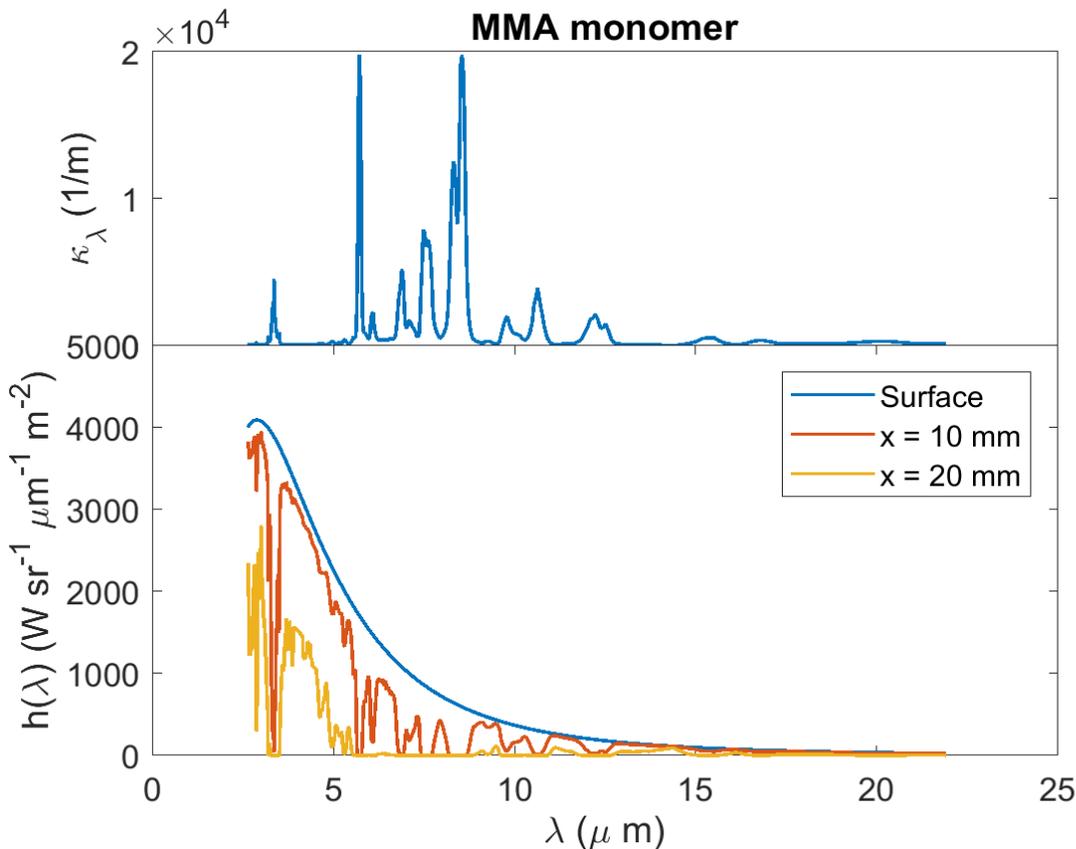


**Pair discussion:**

**Draft the shape of the spectral emitted intensity at different depths when  $\kappa = 1 \text{ m}^{-1}$  around  $\lambda = 0.7 \text{ }\mu\text{m}$  and  $\kappa = 0.1 \text{ m}^{-1}$  elsewhere.**

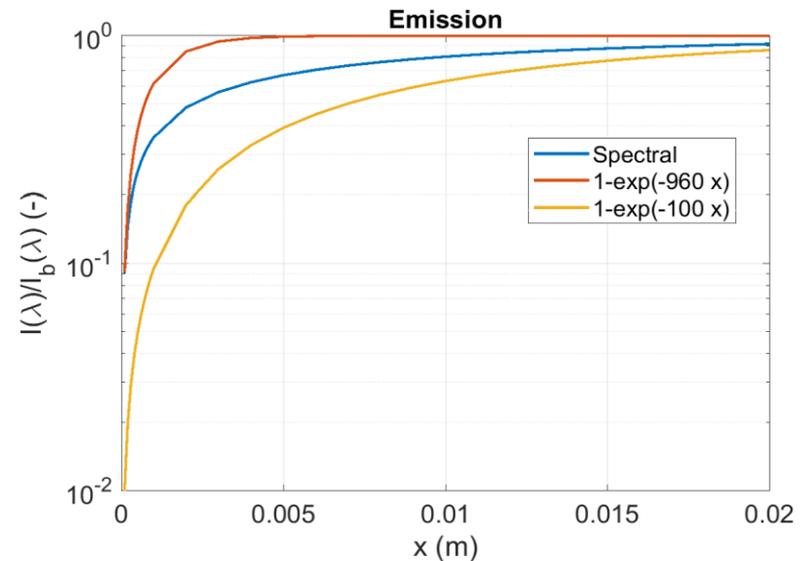
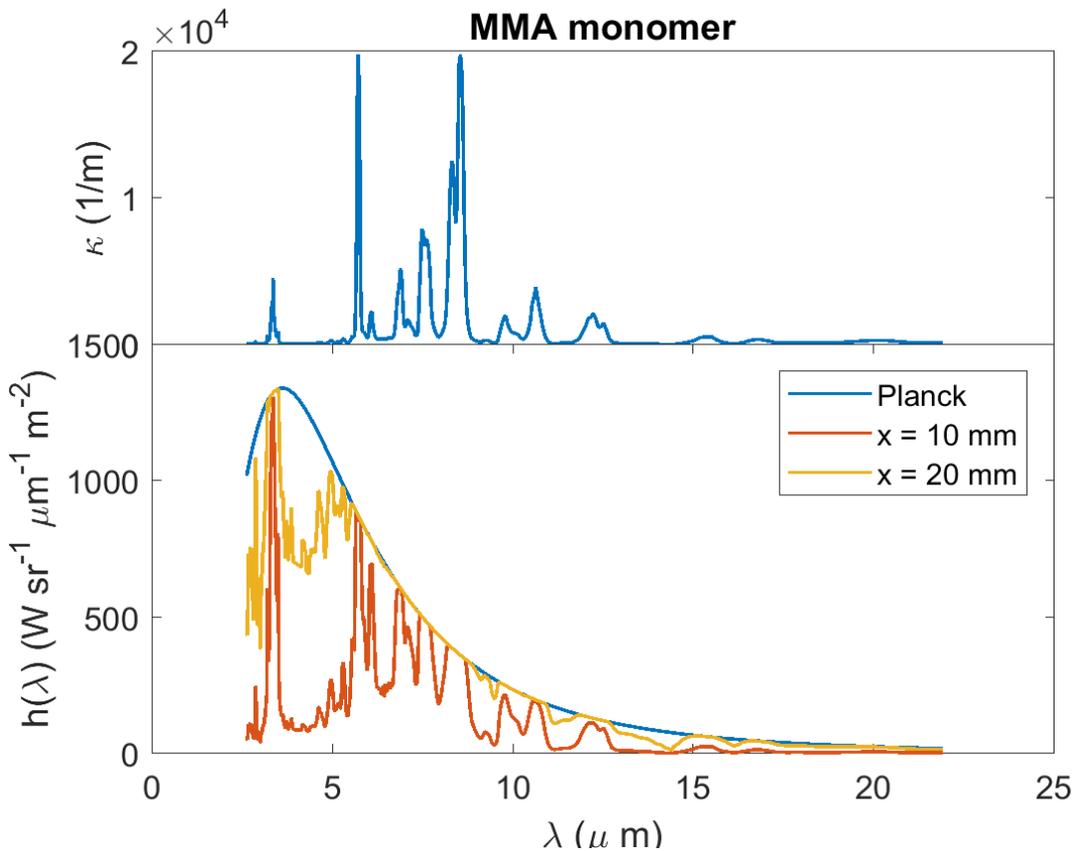
# Example 1: Absorption by MMA

- Blackbody radiation from 1000 K source is absorbed by a cold (0 K) layer of MMA monomer solution.
- Different wavelengths are removed at different distances.



# Example 2: Emission by MMA

- A layer of hot (800 K) MMA solution emits radiation.
- Different wavelengths approach Planck function at different distances.
- There is no average coefficient that would yield a correct profile.



# RADCAL – Narrow Band Model

NIST Technical Note 1402

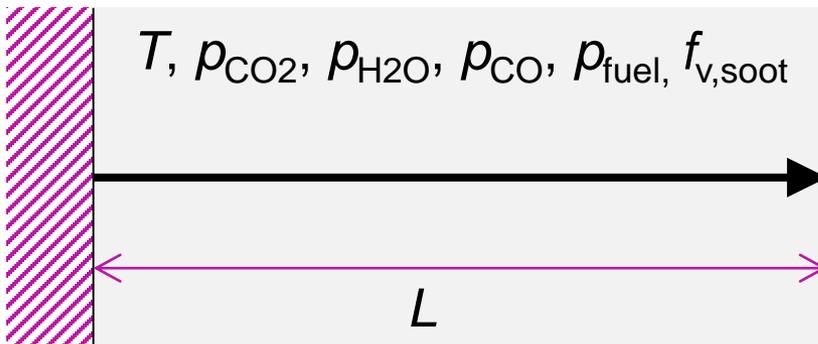
- RADCAL solves the single beam calculation for a large number of wavelength bands and integrates the result over spectrum.
- Outputs: Planck  $\kappa_P$  and effective  $\kappa_e$  mean absorption coefficients.

**RADCAL: A Narrow-Band Model for Radiation Calculations in a Combustion Environment**

William L. Grosshandler

Fire Science Division  
Building and Fire Research Laboratory  
National Institute of Standards and Technology  
Gaithersburg, MD 20899

April 1993



$$\kappa_P = \frac{\int_0^{\infty} I_{b,\lambda} \kappa_{\lambda} d\lambda}{\int_0^{\infty} I_{b,\lambda} d\lambda}$$

$$I_{\lambda}(L) = I_{\lambda,w} e^{-a_{\lambda}L} \int_0^{\tau(L)} I_{b,\lambda} \exp[-(\tau(L) - \tau(L'))] d\tau(L')$$

$$I(L) = \frac{\sigma}{\pi} \left[ e^{-\kappa_e L} T_{rad}^4 + (1 - e^{-\kappa_e L}) T^4 \right]$$

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# Optical properties of gases

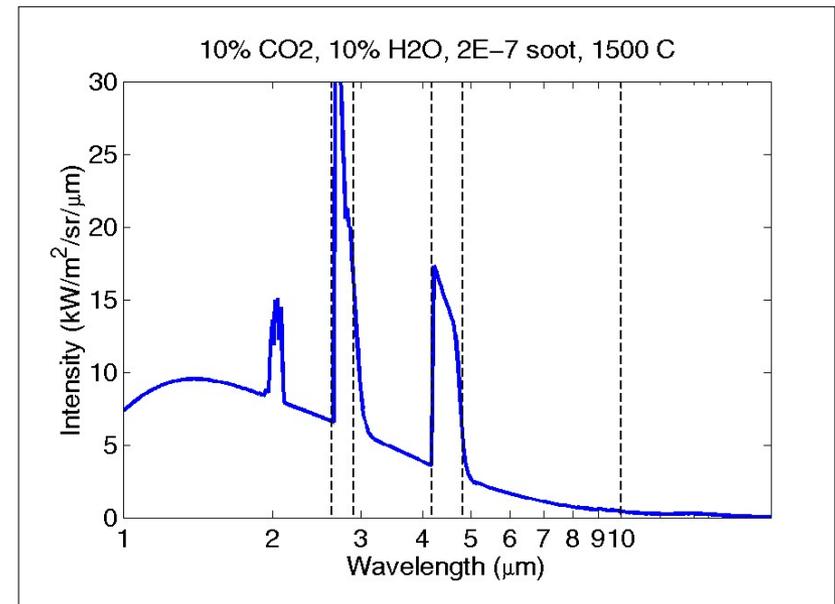
- Original RADCAL included CO<sub>2</sub>, H<sub>2</sub>O, CO and CH<sub>4</sub> –bands + soot.
- Other fuels were added to FDS by Vivien Lecoustre.
- Non-gray model is called wide band model

METHANE,  
ETHYLENE,  
ETHANE,  
PROPANE,  
N-HEPTANE,  
METHANOL,  
TOLUENE,  
PROPYLENE,  
MMA

CARBON DIOXIDE,  
CARBON MONOXIDE,  
WATER VAPOR,  
SOOT

## FDS User's Guide:

RADCAL\_ID  
FUEL\_RADCAL\_ID  
WIDE\_BAND\_MODEL



# Soot absorption coefficient

RADCAL calculates the soot absorption coefficient as a product of volume fraction  $f_v$  and specific absorption coefficient  $\kappa'_\lambda$ .

$$\kappa_\lambda(\mathbf{x}) = f_v(\mathbf{x}) \kappa'_\lambda = f_v(\mathbf{x}) \frac{7}{\lambda}$$

W. H. Dalzell and A. F. Sarofim **Optical Constants of Soot and Their Application to Heat-Flux Calculations.** *J. Heat Transfer* 91(1), 100-104 (1969) doi:10.1115/1.3580063

# Gas property calculation procedure

## **INIT RADIATION \***

```
LOOP over T (K), concentrations (J), and species (NS)
  Call RADCAL(AP0,AMEAN)
  RADCAL_SPECIES2KAPPA(NS,J,K) = MIN(AMEAN,AP0)
End Loop
```

## **SOLVE RADIATION**

```
For each location (I,J,K)
  KAPPA_GAS(I,J,K)=KAPPA_GAS(I,J,K)+GET_KAPPA(ZZ_GET, TYY)
End for
```

## **FUNCTION GET\_KAPPA(Z\_IN, TYY, IBND)**

```
DO N = 1, N_RADCAL_ARRAY_SIZE
  GET_KAPPA = GET_KAPPA + RADCAL_SPECIES2KAPPA(N,Z_IN, TYY)
ENDDO
```

# Source term of RTE

- The emission power density from gas mixture

$$B(\mathbf{x}) = \int_{V_{ijk}} \kappa(\mathbf{x}) I_b(\mathbf{x}) d\mathbf{x} \approx \int_{V_{ijk}} \kappa(\mathbf{x}) d\mathbf{x} \times \frac{\sigma}{\pi} \int_{V_{ijk}} T(\mathbf{x})^4 d\mathbf{x} \approx \bar{\kappa}_{ijk} \frac{\sigma}{\pi} \bar{T}_{ijk}^4$$

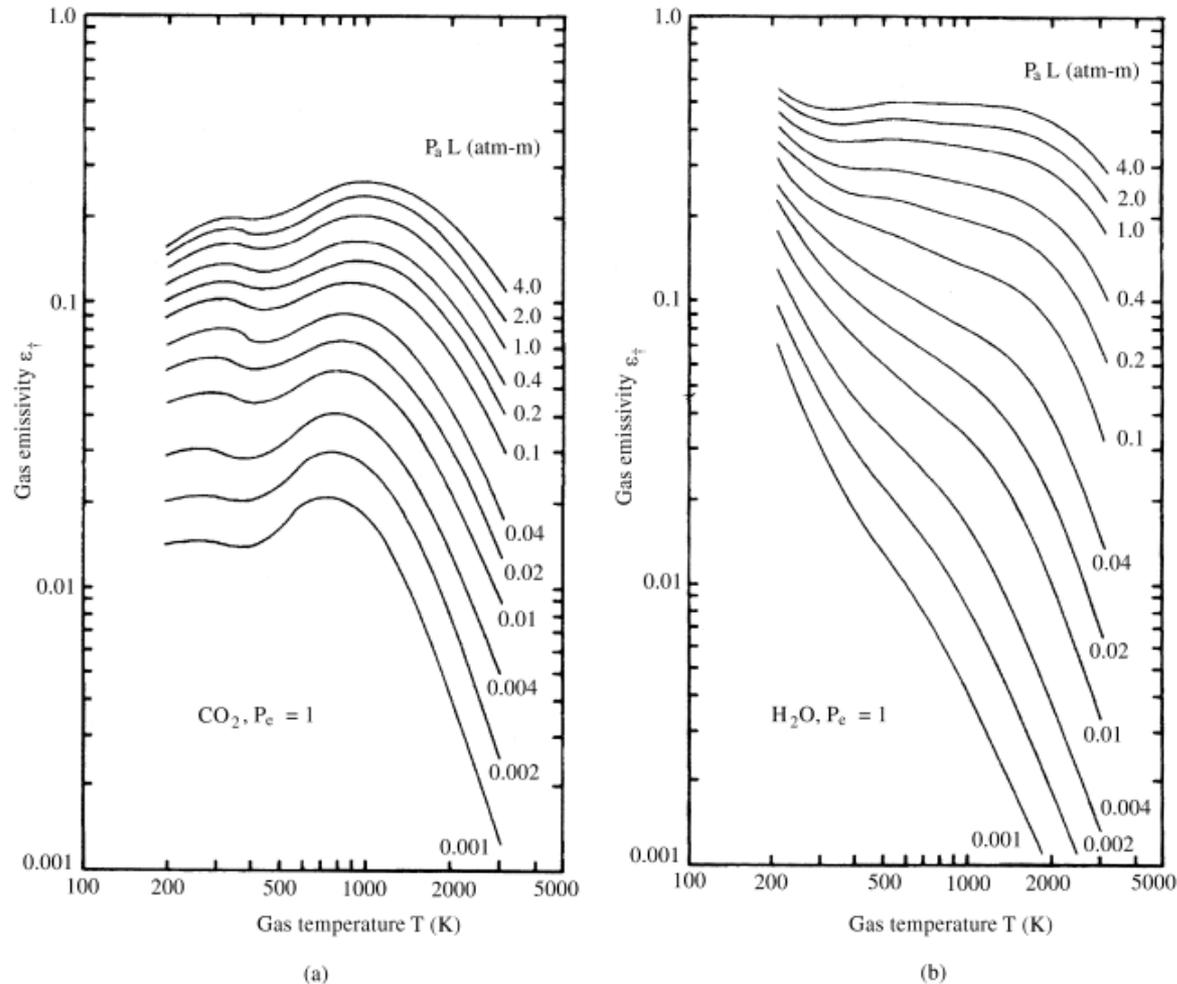
- Three error types:
  - 1)  $\kappa$ - $T$  correlations (Radiation-Turbulence Interaction)
  - 2) T-distribution with a cell  $\langle T^4 \rangle \neq \langle T \rangle^4$
  - 3)  $\kappa$  and  $T$  errors, e.g.  $(0.85 T)^4 \approx 0.52 T^4$
- Usually we want to ensure a radiative fraction  $\chi_r q''' \Rightarrow$  Modelled

$$I_{b,f}(\mathbf{x}) = C \frac{\sigma T(\mathbf{x})^4}{\pi} \quad ; \quad C = \frac{\sum_{\dot{q}'''_{ijk} > 0} (\chi_r \dot{q}'''_{ijk} + \kappa_{ijk} U_{ijk}) dV}{\sum_{\dot{q}'''_{ijk} > 0} (4 \kappa_{ijk} \sigma T_{ijk}^4) dV}$$



# Exercise 2: Gas emission

# Exercise 2: Hottel's charts

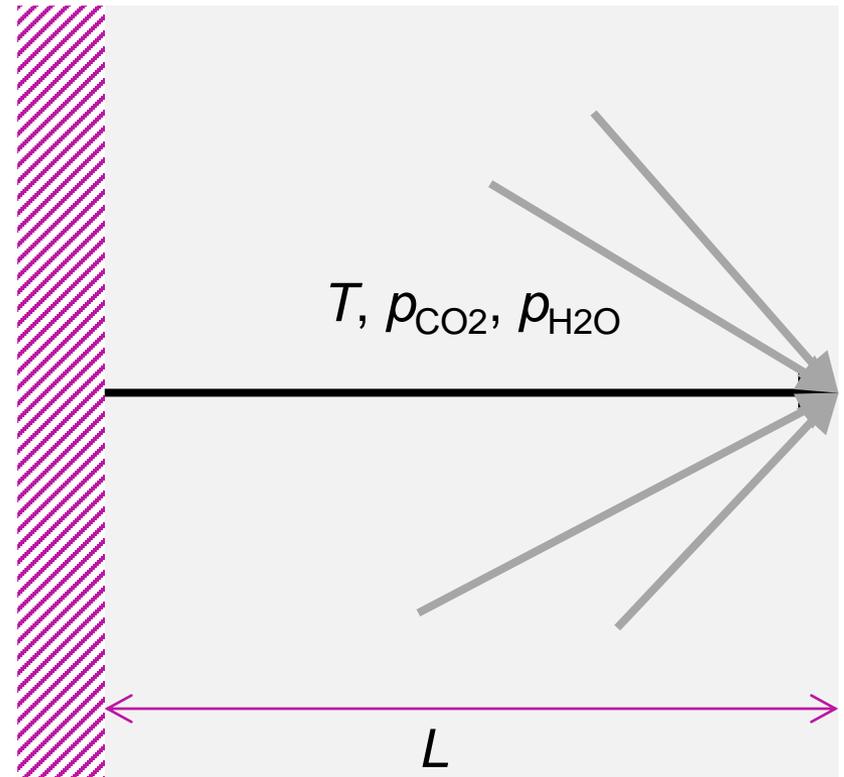


**Figure 2.32** (a) Emissivity of carbon dioxide at 1 atmosphere total pressure and near zero partial pressure. (b) Emissivity of water vapour at 1 atmosphere total pressure and near zero partial pressure. From Edwards (1985). Reproduced by permission of the Society of Fire Protection Engineers

# Exercise 2: Emissivity calculation

- Classical beam emissivity would come from  $I(L) = \underbrace{(1 - e^{-kL})}_{\varepsilon} \frac{\sigma T^4}{\pi}$
- Here, we assume that the Hottel's 'gray body' behaves as a layer, and the emissivity becomes

$$\varepsilon = \frac{\mathbf{q}_{incident}(L)}{\sigma T^4}$$



# Exercise 2: Reproduce Hottel's curve

1. Study the input file **hottel\_array.fds**
2. Pick one of H<sub>2</sub>O or CO<sub>2</sub> curves in Hottel's charts, and specify the corresponding gas concentration into the FDS model.
  - Hottel's curves use product  $\rho_{\text{gas}}L$
  - In the model,  $L = 1$  m and  $P_{\text{tot}} = 1$  atm  $\Rightarrow \rho_{\text{gas}}L = \text{volume fraction}$ . So, you just need to specify volume fraction. But FDS initialization only works for mass fraction. You calculate them as

$$Y_{\text{CO}_2} = \frac{X_{\text{CO}_2} 44}{X_{\text{CO}_2} 44 + (1 - X_{\text{CO}_2}) 28.8} \quad Y_{\text{H}_2\text{O}} = \frac{X_{\text{H}_2\text{O}} 18}{X_{\text{H}_2\text{O}} 18 + (1 - X_{\text{H}_2\text{O}}) 28.8}$$

- CO<sub>2</sub> mass fraction can be specified using `Y_CO2_INFTY` at `MISC` line.
  - H<sub>2</sub>O mass fraction must be set on `&INIT` lines
3. Run the model and plot the emissivity at different temperatures using **plot\_hottel.py**.
  4. Discuss with your neighbor how the code works.